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First Line Comprehensive Care. Part II: Anthropogenic Xenobiotics in Functional Medicine. Managing Persisting Bioaccumulating Pollutants: Toxic Minerals, Biocides, Hormone Mimics, Solvents, and Chemical Disruptors

Russell Jaffe, MD, PhD, CCN, NACB

First line comprehensive care is fundamental to integrative medicine, an emerging specialty within American healthcare. Primary focus is on functional and predictive tests to identify remediable causes of suffering and ill health. Integrative health professionals are learning a more functional language—it is both a language of causes and of deeper insights into more fundamental molecular and submolecular mechanisms of both our good health and our ill health. Anthropogenic xenobiotics, especially the bioaccumulating and bioconcentrating toxic minerals (TMs) and persistent organic pollutants (POPs) are the focus of this article as they are, in their more toxic forms, largely anthropogenic, human-sourced intoxicants. Bioconcentration operates in the following way: Ocean krill and algae convert less toxic, inorganic mercury to biotoxic organic mercury. Algae are ingested by small fish. Larger fish eat small fish, and humans, in turn, eat larger fish. The longer the lifespan, the more mercury accumulates in those people who have lost (phenotypic expression) or have innately impaired (genotypic expression) abilities to detoxify and eliminate these anthropogenic xenobiotic toxicants. Since accumulation of POPs takes place primarily in the fat while TM accumulates more prominently in muscle and bone cells and extracellular matrix, it follows that such toxins might contribute to the novel and more severe treatment-resistant musculoskeletal conditions observed in practice. Scientific evidence is also “bioaccumulating” for a clinical imperative to mitigate exposure where possible and to optimize innate host defenses, generally through strategic mental and functional, nutritional, and environmental adaptations. This article outlines the current standards of care for integrative medicine physicians incorporating the 2005 American Board of Clinical Metal Toxicology (ABCMT) training guidelines for practitioners. The practice of first line comprehensive care is emerging as outcome and cost effective compared with today’s conventions in most chronic diseases as well as in the implementation of proactive, cost-effective, and outcome-effective health promotion approaches. If not now, when will we implement widely what we know that cost effectively and outcome effectively promotes sustainable good health and general well being?

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Russell Jaffe is Senior Fellow of the Health Studies Collegium research foundation. He is developer of various functional methods from occult blood detection in early colon cancer screening to platelet dose–response aggregation studies in coagulation to the LRA by ELISA/ACT tests and plans, the D-penicillamine nutritional and toxic mineral assessment protocol, the ascorbate calibration protocol, magnesium uptake protocol, glutamine recycling protocol, among others. He serves as director of ELISA/ACT Biotechnologies, LLC and, PERQUE, LLC, nutritive supplements.

He is recipient of the International Scientist of the Year 2003 awarded by the International Biographical Commission of Cambridge, England to recognize his contributions to Biochemistry, Clinical Medicine, and Immunomics. He was recently elected as a Fellow of the National Academy of Clinical Biochemistry. He also maintains Fellow status in the American Society for Clinical Pathology, American College of Nutrition, American College for Allergy, Asthma, and Immunology, American Medical Laboratory Immunology, and the Federation of Clinical Immunology Societies. He is also a certified clinical nutritionist (CCN).

Address reprint requests to Russell Jaffe, MD, PhD, CCN, NACB, 14 Pidgeon Hill #300, Sterling, VA 20165. E-mail: Rjaffe@rmjholdings.com

First line comprehensive care is an increasingly sought choice by patients, clinicians, and frontier scientists, looking functionally for causes; recognizing the role of the integration of mind and body; appreciating our adaptation to our environment; using predictive and provocative labs tests and procedures that are more sensitive, specific, and reliable than usual conventional tests. This article focuses on the approach taken by integrative medicine to manage anthropogenic xenobiotic toxins and clinically useful antitoxin strategies and tactics.

Xenobiotics are biologically active synthetic chemicals, many of which compromise human health. The public health burden due to anthropogenic xenobiotics, which come primarily from toxic minerals (TMs) and persisting organic pollutants (POPs), is an acquired and reversible health tax for over 200 million Americans. According to the Environmental Protection Agency (EPA), over 104,000 synthetic chemicals have been introduced into the environment in the past century. Barely 4,000 (about 4%) have been studied for their toxicities. Only a handful have been studied for their interactions.

The human cost is a reduction of 8.8 years of life for the average person due to the effects of these toxicants.¹ This is a biological tax of 10% of most people's life span. The direct disease care costs induced by TMs are calculated, in aggregate, to be in excess of \$100 billion annually (EPA, 2000; Princeton University, 2001).² The public health risk from TMs is even yet greater due to observed but not extensively defined or replicated synergies of mineral toxicities.³ The direct disease care costs induced by POPs are less precisely known yet are about the same cost as TMs.

As Nriagu and Pacyna⁴ remind us, "our toxic metal time bomb's impact on human health and on our ecosystem can be compared in importance to the total radioactive waste in need of disposal or the excess carbon dioxide production associated with enhanced greenhouse gas effects."⁵ This is understandable given that there has been a 1,000-fold increase in TMs and POPs in our environment over the last 1,000 years, most of it in the last century. Bioaccumulation in mammals, including humans, is typically 100,000 to 200,000,000 times that of the environment. A little TM and POP can go a long way to impact human health in the 21st century.

This is largely due to most mammals' ready uptake and increasingly impaired innate release (detoxification + excretion) mechanisms when they lose innate antitoxic, homeostatic mechanisms. These innate mechanisms are designed to trap and facilitate the safer elimination of these toxins. Further, these mechanisms are inducible when we come in contact with small amounts of the toxicant and have the healthy resilience to induce elective, protective mechanisms.

In contrast, today, too many people have lost those protective mechanisms and thus appear to be at greater (genetic) risk that is actually acquired (phenotypic). Functional tests are necessary to separate these differences.

Still further, we know all too little about interactions of toxins. What little we know suggests there may be profound synergies. We are learning too late that an uncontrolled ex-

periment on the entire population is a difficult one to control. Since this experiment started before informed consents were routinely required, it is understandable that most of us have not signed a consent to participate in this anthropogenic, xenobiotic experiment in adaptation and response to multiple toxic burdens.

This review covers the impact of well-studied, clinically known toxicant groups on chronic conditions. In addition, functional tests to determine body toxicant burden and immunotoxic reactivity are included because they improve diagnostic precision and clinical outcomes. Functional procedures such as penicillamine provocation for nutritional and TM status allow a noninvasive clinical window on cellular mineral and cellular buffering competencies. Analysis of fat biopsies or of postsauna blood levels for POPs can be performed following EPA criteria methods. Still further, lymphocyte response assays (LRA) by the ELISA/ACT method, late-phase, delayed hypersensitivity reactivities allows for patient-specific diagnostic testing and therapeutic monitoring. This gives us information on the presence or absence of homeostasis and tolerance. Specific reactive items thus identified can be substituted, reducing the defensive burden on the immune system.

Taken together, these approaches to clinical management offer more predictive, cost-effective, outcome-effective, integrative, and comprehensive clinical care.

Persistent Organic Pollutants

POPs are a broad category of synthetic chemicals including PCBs, dioxin, chlordane, and DDT. POPs are pervasive chemicals. They can be categorized as follows.

1. Hormone disruptor biocides (pesticides, fungicides, mitocides)
 - a. Cholinesterase inhibitor organophosphate pesticides
 - b. Halogenated pesticides
2. Solvent residues
 - a. Chlorinated compounds (chloroform, methylene chloride, ethylene chloride . . .)
 - b. Other halogenated compounds (brominated, flourinated, iodinated) used most commonly as artificial food dye colorants, radiocontrast agents, and art materials.

Even though some POPs have been banned or restricted in use by some countries, POPs are, as their name suggests, persistent in the environment. This means they bioaccumulate. Designed to stability, they persist and accumulate in the environment. They evaporate slowly into the atmosphere and disperse around the globe. Living organisms then concentrate these fat-soluble chemicals in fatty tissues.

Detection of POPs

Adverse effects to human health can begin at thresholds below direct detection. In the case of dioxin, PCB, PBB, and related compounds, human health risks emerge at the parts

per trillion (ppt) level. This is in contrast to most lab tests that are only able to measure down to parts per million (ppm) or parts per billion (ppb) levels of detection. In other words, we now routinely have biological health effects at amounts of materials in our bodies below our ability to detect them. From chronic fatigue (CFIDS) to endometriosis, from adult muscular dystrophy (inclusion body myositis; IBM) to reactive depression, increasingly, treatment-resistant conditions are consuming our resources without improving our health or productivity.

Toxic Metals

While TMs with balanced electrons, such as metallic lead or mercury, are of low direct human toxicity,⁶ their surprisingly ready conversion under physiological conditions to substantially more toxic biologically active forms (eg, divalent methylmercury, dimethylmercury, mercuric sulfides, and other mercurous/mercuric compounds, ethyl-lead, etc) continues to be a major public health risk.⁷ Biologically active TMs are considered by Nriagu and colleagues⁴ to be the most toxic of all the toxic anthropogenic exposures in the biosphere even compared with POPs and ionizing, radioactive elements.

The common TMs encountered in North America are lead (Pb^{2+}), mercury (Hg^{2+}), arsenic (As^{2+}), cadmium (Cd^{2+}), nickel (Ni^{2+}), and aluminum (Al^{3+}). Except for trivalent aluminum and ferric iron along with exotic multivalent minerals, TMs are divalent. This predicts their transition state biochemistry.

Primary sources of TM exposure in humans include

1. Medications and devices as well as amalgams and vaccines
2. Occupational or recreational settings
3. Fungicides and mitocides in interior environments (eg, paints or fumigations) or exterior environments (eg, agriculture or gardening)
4. Recreational exposures including from leaded glass decanters for beverages or ceramics used for food, ceramic glazes used by artists and commercial glazers, commercial uses of solders and fluxes
5. Water and aerosol contamination
6. Dietary sources including fish, fowl such as commercially raised chicken, beef and game; generally, the higher up the food chain, the greater the bioconcentration of contaminants

Living in an industrialized society exposes all inhabitants to metals and POPs in the environment. Substantial sources of highly toxic compounds can greatly enrich an environment in a toxicant without public awareness. These largely invisible depositions are bioidentical and just as toxic as the exposures of which we are aware. For example, 200–600 tons of mercurial toxicants are annually added to the American ecosystem from all anthropogenic sources. An additional 100 tons of mercury is derived from trans-Atlantic tiny dust particles. Additional metric tons may be added by the trans-Pacific plumes of aerosol toxicants by our Pacific Rim neighbors. These are carried in the upper atmosphere and contain

enough mercury and arsenic to qualify as mineable ore if only this dust could be trapped before it reaches the southern United States and Caribbean Basin (Seba D, personal communications, 2000–2001).

This last environmental burden was unknown until as recently as 1990. This illustrates how substantial sources of “high toxic effects compounds” can greatly enrich an environment in a toxicant without general awareness of the influx of that toxicant. These largely invisible depositions are bioidentical and just as toxic as the exposures of which we are aware.

For example, the 300 tons of mercury added to the American ecosystem annually, when expressed in micrograms is 1.18×10^{15} . Given that TM effects are usually measured in micrograms, there are, then, about a quadrillion toxic doses of mercury released into the environment each year. With a population of 300 million (or 3×10^8) in the United States, this equates to $3.93 \times 10^7 \mu g$ (39,000,000 μg) per citizen per year from anthropogenic, manmade sources.

Toxic metals are potent metabolic, hormonal, immune, and gene toxins.⁸ By example, continued exposures to TMs when they bioaccumulate above about 1 ppb^{9,10} impose long-term human health risks, particularly as treatment-resistant increased chronic autoimmune and cardiovascular illnesses.¹¹

With regard to lead, the evidence base of pervasive subacute toxicities is particularly well documented and well reviewed elsewhere.¹² A prudent person would extend what we have learned about the insidious yet deep toxicities of lead to other TMs with similar electrochemistry.

In the case of dioxin, PCB/PBB, and related compounds, human health risks for these POPs emerge at the parts per trillion level. This is in contrast to most lab tests that are only able to measure down to parts per million levels of detection. In other words, we now routinely have biological health effects at amounts of materials in our bodies below our ability to detect them.

Living in an industrialized society exposes all inhabitants to TMs and POPs of no human benefit. In contrast, some minerals are essential for life. To some extent, beneficial minerals are antitoxic in that they block or compete with the TMs for uptake or action in cells. Said another way, people with adequate stores of buffering minerals block the uptake and facilitate the excretion of TMs from the body in all excretory pathways. These pathways of excretion include urine, stool, sweat, desquamated skin, hair, nails, and breath. The antitoxic minerals work best when the amounts, balance, and ratios provide ample reserve pools to draw upon. The particular minerals involved are potassium and sodium, calcium and magnesium, zinc and copper, chromium and vanadium, manganese and molybdenum, selenomethionine, and both iodine and iodides (Lugol’s solution).

Some bioactive minerals, such as selenium (selenomethionine and selenocysteine), can form stable, permanent, covalent links with biologically active mercury or arsenic (and, probably, other divalent TMs), thereby detoxifying them. These stable complexes are not easy to remove and may remain in the body for periods of years to decades. Their “bal-

anced electron” relatively low toxicity reduces the priority placed on their removal from the organism. In contrast, other forms of selenium commonly used in supplements (selenite and selenate) do not have these beneficial properties yet are more toxic.

The public health burden due to TMs and POPs is an acquired and reversible health risk for over 80 million Americans. The human cost is a reduction of 8.8 years of life for the average person due to the effects of these toxicants.¹ This is a biological tax of 10% of most people’s life span. The direct disease care costs induced by TMs are calculated, in aggregate, to be in excess of \$100 billion annually.²

This means that we could fund out of savings from sick care costs *not* incurred the transition from our current symptom reactive, sick care focus to a proactive, intoxication prevention program. The public health risk from TMs is yet even greater due to observed but not extensively defined or replicated synergies of mineral toxicities.³

Since TMs and POPs both bioaccumulate and bioconcentrate, the adventitious exposures are likely to increased greatly just at the time that internal reserve mineral and antioxidant protectors are at their highest. It is important to address the alkaline balance and avoid both metabolic acidosis and catabolic illness.

Toxic Minerals: Free Radical Damage That Potentiates Chronic Diseases Especially When Deficits in Antioxidant and Buffering Minerals Are Present

TMs and POPs are amongst “nature’s mimics” in that they can bring to substantially greater symptomatic intensity a wide variety of clinical conditions. From cardiovascular diseases to fibromyalgia and rhabdomyolysis chronic pain, from birth defects to autism, from premature senility to autoimmune syndromes, from endometriosis to CFIDS, TMs and POPs potentiate these highly diverse conditions due to their common yet variable locus of expressions of free radical pathology, made possible by relative antioxidant deficits and metabolic acidosis.

Understanding this molecular pathophysiology allows us to enter a new era of clinical medicine in which comprehensive, integrative care plays an important role. Identifying the role of clinical chronic subacute (low level, yet persisting) TMs and POPs actions is integral to the “identify and mitigate the causes rather than focus on relief from the symptomatic consequences” philosophy of care that distinguishes integrative and comprehensive care as a specialty from conventional internal medicine, family practice, and their subspecialties.

Among the effects of TMs and POPs are the following molecular consequences for cell functions:

1. Metabolic uncoupler: Inhibits cytochrome I to coenzyme Q10 transfer in the mitochondria, the cell’s

energy-producing, detoxifying saprophytic organelle. This reduces ATP (high energy compound) production thus reducing the functionality of those parts of the cell consuming the most energy; generally this means the most metabolically active and functionally important component of the cell becomes starved for energy. This promotes fatiguability and reduces elective, proactive defenses along with immune competencies.

2. Hapten immunotoxins: Small molecules that bind to and distort the structure of the body’s own proteins, from globulins to insulin, from lipoproteins to macroglobulins, thus increasing the probability of autoimmune, chronic illness.
3. Enzyme inhibitor: For cell regulatory control kinases and other enzymes with cysteine or thiamine (B1) sulfhydryls at their active site. Phosphodiesterase, superoxide dismutase (SOD), and nucleotide binding protein (NBP) are examples of particularly vulnerable, functionally vital enzyme catalysts.
4. Anti-antioxidants (prooxidants): TMs and POPs are prooxidants that induce excessive consumption and wasting of glutathione and ascorbate. These two antioxidants are at the center of the antioxidant recycling network that protects delicate cell components from free radical oxidative damage during physiologic homeostasis and immune tolerance. Once antioxidants are selectively depleted, free radical damage can “run rampant” in oxidizing and making dysfunctional important cell systems.
5. Bioconcentrate: These toxicants bioaccumulate in critical and most unfortunate places within the body, such as the choroids plexus, where spinal fluid is produced, to the loop of Henle, where the kidney concentrates toxins for excretion into the urine. This reduces the body’s functionality and accelerates biological aging.

TMs and POPs effects are more destructive when low cellular minerals (particularly potassium and magnesium) dispose the cell to metabolic acidosis, however well compensated the blood pH may be. The combination of metabolic acidosis and TMs/POPs, for example, accelerates mineral leaching from bone to buffer the excess acids the kidney needs to concentrate and excrete without damaging itself, to the extent possible. This means that osteopenia and osteoporosis are accelerated. Among the other effects are shifts within cells from elective production of structural proteins and metallothioneins to a survival mode for the cell such that only actions needed for the cell to avoid death are performed.

Elective and protective elements are no longer produced under the above conditions. Toxic effects of TMs are further potentiated in this situation. Unhealthy hormone metabolites may accumulate rather than be excreted due to toxic damage or lack of cell energy required to “pump” toxicants out of the body and into urine, stool, and sweat, while healthier hormone products, in contrast, may not be made. Further, we know too little about the interactions of low-level persistent TMs. What little we do know suggests that there may be a synergy (rather than addition) of toxic effects when more

than one TM is concurrently present (ATSDR and EPA, personal communications, Fowler B, Sonawane B, 2001).

Absence of Evidence Is Not Evidence of Absence

With regard to POPs and possible interactions among chemicals, of the 104,000 chemicals introduced in to the environment through manmade novel synthesis, barely 4,000 have been studied at all and barely a handful have been studied for their interactive toxicities. The absence of data is not a basis for assuming the data of absence exists (Sonawane B, personal communication, 2004).

For example, the EPA recommended in 2001 a 10-ppb arsenic maximum acceptable level in drinking water.¹³ The Institute of Medicine of the United States National Academy of Sciences expert panel on arsenic recommends a drinking water standard of less than 1 ppb because the cancer-promoting effects of even this level of arsenic in the water are deemed to be too high.¹⁴

Arsenic, at 1 ppb in the drinking water, increases the risk of cancer by 1 in 1,000 in a lifetime. As toxicologists, we are used to thinking about risks in terms of excess cancers per million people exposed. Thus 1 ppb arsenic in the drinking water over a lifetime increases the risk of cancer by 1,000 per 1,000,000 people. This is above the historically accepted, conservative EPA risk threshold of one extra cancer per million population. To many physicians and scientists, even this level of risk is unacceptable given that cost effective “mitigation at the source” solutions are available. Examples of this approach are given in the recent book *Natural Capitalism*.¹⁵ Other examples are the report of to the Department of Consumer Affairs of the State of California titled *Clean Your Room*.¹⁶

Other examples of absence of evidence is the immune system’s response to toxins, even once removed. There are T-helper lymphocytes that are involved with delayed allergy reactions to haptenic^a immunotoxins such as toxic minerals. Clinically, this can be functionally measured by a classic Memory Enzyme Linked ImmunoSorbent Assay (MELISA) modification of thymidine incorporation or by the LRA by ELISA/ACT tests that assays kinase activation prior to inducing thymidine incorporation and cell mitotic division. These technologies show us that even tiny amounts of internal or environmental exposure to a substance that induces an immunotoxic hypersensitivity reduces immune defense abilities and induces deferral of immune repair. When this becomes the “usual condition,” we become hospitable to “whatever is going around.”

Further examples still of evidence’s absence is the toxin’s damage to energy production. Among the effects of TMs and POPs when they bioaccumulate in cells is apoptosis of mito-

chondria. After mitochondria commit this programmed cell suicide, the cell’s overall death is not far off. In contrast, protection and rehabilitation of mitochondria is central to lifelong health maintenance, restoration, or enhancement.

TMs and POPs effects are more destructive when low cellular minerals (particularly potassium and magnesium) predispose the cell to metabolic acidosis. Intracellular depletion of potassium and magnesium can cause metabolic acidosis regardless of how well compensated the blood pH may be. The combination of metabolic acidosis and TMs and POPs accelerates mineral leaching from bone to buffer the excess acids the kidney needs to concentrate and excrete without damaging itself, to the extent possible. This means that osteopenia and osteoporosis are accelerated. Among the other effects are shifts within cells from elective production of structural proteins and metallothionines to a survival mode for the cell such that only actions needed for the cell to avoid death are performed. This net acid excess (NAE) can be clinically assessed through measurement of first morning urine pH (after 6 or more hours of rest; a healthy pH is 6.5-7.5; <6.5 is presumptive evidence of cellular metabolic acidosis).

Elective and protective elements are no longer produced when buffering minerals and protective antioxidants are in deficit. Effects of TMs are further potentiated. Unhealthy hormone metabolites may accumulate rather than be excreted due to toxic damage or lack of cell energy required to methylate or otherwise detoxify and “pump” toxicants out of the body and into urine, stool, and sweat while healthier hormone products, in contrast, may not be made. Further, we know too little about the interactions of low-level persistent TMs. What little we do know suggests there is synergy of toxic effects when two or more TMs are concurrently present (ATSDR and EPA, personal communications, Fowler B, Sonawane B, 2001).

Risk of Bioaccumulation of Toxicants: A Matter of Balance and Homeostasis

Bioaccumulation of TMs or POPs is a function of intake and output (I/O) balance.

$$\text{Intake} - \text{Output} = \text{Residual (remainder in body)}$$

The integral of this simple input–output model determines individual body burden. For example, if

1. Intake (high) – output (low) = Increase in toxic burden
2. Intake (high) – output (high) = Steady state, high risk state
3. Intake (low) – output (low) = Decrease in toxic burden
4. Intake (low) – output (low) = Low exposure

Goal state:

Intake (low) – output (high) = Body burden reduction

It is possible to reduce, but not avoid, intake as discussed earlier. It is also possible to increase protective output to

^aHaptenic substances (or haptens) are small molecules, which, while not large enough to be recognized as foreign by the body, bind to the body’s own proteins. This binding distorts the innate structure rendering them “foreign” and immunoreactive in the body.

reduce the residual, or total body burden. In other words, low intake and high capacity to excrete toxins is a feasible clinical evidence-based goal.

Note that high output is associated with elective synthesis of metallothionines, polypeptides made largely of glycine and cysteine with zinc or magnesium as the counter ion. When these biological detoxifiers are produced, the exhibit substantial TM trapping capacity if observed in the gut, the plasma, and the cerebrospinal fluid.

Increasing Output to Reduce Bioaccumulation

William Walsh of the Pfeiffer Treatment Center, Wheaton, Illinois, suggests a hereditary or xenobiotic pseudogenetic predisposition to mercury toxicity and/or T lymphocyte hypersensitivity (DTH).¹⁷ These emerging data makes thimerisol exposure at times of distress or impaired detoxification particularly troublesome. Thimerisol typically contains 5-7.5 μg of ethylmercury per vaccination dose.

High output of toxins from the human body is associated with elective synthesis of metallothionines. Under normal circumstances, there is a large concentration of the protein metallothionine waiting in the intestines, as a sentinel, to interact with the mercury or other toxic mineral and detoxify it before it enters the body.

Each metallothionine molecule has binding sites for seven atoms of zinc plus variable amounts of magnesium, selenomethionine, and glutathione. Structurally, it is a linear protein of 61 amino acids with 20 or more cysteine or cystine active sulfhydryls. Its job is TM detoxification. It is present in high concentration in the gastrointestinal tract and in the liver, but it is present in every cell in the body. When present, it protects the gastrointestinal tract from all of the nasty things that TMs like mercury can do. However, its production is elective. Metallothionine production occurs *only* when the body is healthy, alkaline buffered, and in homeostatic equilibrium. In states of hormonal, neurochemical, or immune distress, metallothionine production can be substantially down-regulated.

If you take somebody whose metallothionine system is working well, however, the mercury forms covalent links to other, active sulfhydryl groups. The sulfhydryl groups in the active site of certain enzymes in the gastrointestinal tract include the enzymes that break down casein from cow milk and gluten from wheat and other grains. A metallothionine disorder, therefore, is often associated with major digestive and/or dysbiosis problems as well. Most typically, wheat and casein intolerances and other delayed T cell—mediated allergic hypersensitivities occur. These individuals are also prone to intestinal inflammation and enteropathy.

Metallothionine is a family of four proteins (1, 2, 3, and 4). Metallothionines 1 and 2 are ubiquitous and present in every cell in the body. Metallothionines carry out innate antioxidant functions and/or to deliver zinc

wherever it is needed. (Presentation at the Princeton BioCenter, May 1990)

Metallothionine is also responsible for homeostasis between copper and zinc. These trace elements, in turn, are related to production of specific hormones, cytokines, and neurotransmitters. For example, for the zinc or copper requiring enzyme catalysts to convert the right amount of dopamine into norepinephrine, copper to zinc balance and sufficiency are required.

Walsh and colleagues have used the plasma zinc/copper ratio as an indicator of properly functioning metallothionine. They use it as an indicator of “toxic-coping ability.” They report

1. In a population of obsessive-compulsive individuals, the ratio between serum zinc and copper is, typically, around 0.8.
2. The healthy range, based on 100,000 individuals, is $\sim 1.0 \pm 0.1$.
3. Walsh and coworkers examined 5,700 individuals with attention-deficit disorder; the mean ratio was 1.17.
4. For children who exhibit violent behavior, the ratio is typically > 1.4 .

Walsh suggests that impaired homeostasis for copper and zinc correlates with poor metallothionine function. The detailed influence of supplementation on normalizing these ratios and their impact on function and performance is, as yet, unreported (Walsh W, personal communication, 1999). The anecdotal evidence is encouraging, according to Walsh.

Reducing Iatrogenic Input

If you study people with amalgams, many people show few adverse effects. Similarly, most children who receive vaccinations containing thimerosal go through this experience without many notable adverse effects. Perhaps these are the individuals with adequate ascorbate and glutathione, magnesium and zinc, selenomethionine, and sulfur from dietary sources (including breast milk from mothers whose antitoxic levels are high). These individuals are protected and are at relatively low risk. When zinc, selenomethionine, and magnesium are marginal or deficient, metallothionine loses functionality.¹⁸ Such individuals are sensitive and/or at high risk of adverse consequences of TMs and POPs.

The Swedish experience is the most rigorous and extensive regarding toxicity from dental materials, particularly mercury amalgam. Lindh¹⁹ pioneered research using nuclear probe microscopy for minerals in biomedical analysis.^b Neutrophil (granulocyte) mercury in patients with mercury amalgams who were sick was compared to controls (ie, people

^bNuclear microscopy or PIXE is an advanced analytical tool, which allows for the measurement of trace elements in small objects, such as the nucleus of the neutrophil granulocyte (with a detection limit of 0.5 $\mu\text{g/g}$ dry substance. This is done by bombarding the cells and their organelles with protons (hydrogen atoms). Because each trace element has its own characteristic emission fingerprint, it is then possible to determine the amounts of a particular element in the various regions of the cell.

with mercury amalgams who were not sick). The results showed that the patients who had amalgams and who were sick had detectable mercury in their cells and that the controls did not show bioaccumulation of mercury.

In addition, the concentrations of other elements such as magnesium, calcium, manganese, iron, and zinc were more than one standard deviation *below* healthy ranges in the symptomatic subjects and not in the asymptomatic, otherwise matched controls. Examination of elements in the nucleus showed a maldistribution of zinc, which correlated with the presence of mercury in the nucleus of the neutrophils. There is a typical zinc distribution in the nucleus of the neutrophil granulocyte. In contrast to this normal situation, the patients who had mercury burdens showed an abnormal distribution and an invasion of mercury into delicate nuclear or nucleolar centers. Mercury in the nucleus correlated with the decreased zinc in those areas. Whether mercury caused the mineral aberrations or whether preexisting mineral deficiencies predisposed to mercury remains to be determined.

In summary, by using sensitive probes, Lindh demonstrated the presence of mercury in the cells of patients who had amalgams and who were sick and the absence of mercury above threshold levels in the cells of asymptomatic controls with amalgams.

The majority of metals that are used in dentistry belong to the transition group in the periodic table. A general characteristic of these elements is that they have an uncompleted electron shell, either in the natural or oxidative state. Since electrons always exist in pairs, transition metals form strong complexes with both organic and inorganic ligands. The memory cells are long lived and can be detected in the blood of sensitive individuals, prior to the appearance of objectively documented clinical symptoms.

Stejskal and coworkers¹⁹ elucidated the immune response that mercury may trigger. The research agrees that T lymphocytes play a role in all types of allergic and autoimmune reactions. This makes them evident candidates as markers for metal-induced sensitivity. After contact with an antigen, T and B lymphocytes that are antigen-specific for that substance correlate with inflammatory reactions that lead to cell damage when repair is delayed or blocked. Repeated exposure with the same or a chemically similar cross-reacting antigen will immediately induce a faster, secondary immunological reaction initiated by the memory cells. Cytokine release will activate other cell types and the result is either beneficial for the body when repair is facilitated or, in the case of repair-deficient autoimmune diseases, a pathological consequence.

Human lymphocytes can be stimulated *in vitro* with various foreign substances called mitogens. Mitogens refer to substances that stimulate immune responses from memory carrying lymphocytic white blood cells. The lymphocyte stimulation test has been used for 30 years as routine analysis for evaluation of cellular immunity and clinical immunology, as well as for diagnoses of allergic reactions to medicines, metals, and other substances. Specific stimulation is based on the fact that every person's immune system remembers the antigen that it has previously been programmed to remem-

ber.²⁰ Such a reaction gives rise to memory T and B lymphocyte cells that circulate in the bloodstream and defend, as needed, the individual against foreign substances, including

1. Xenobiotics and other synthetic small molecules (mostly haptens)
2. Partially digested, immunoreactive food digestive remnants
3. Pathogens including bacteria, parasites, viruses, or anything recognized by an individual as foreign to their immune system

Other types of white blood cells are dendritic cells such as monocytes and macrophages, endothelial cells and fibroblasts, and astrocytes and Kupfer cells. These cells perform various functions, such as presentation of processed antigens to naïve unprogrammed lymphocytes and removal of toxic substances, thus they are termed "scavenger" or dendritic cells. They are short lived with a typical life span of 8-12 hours in the body. Tests that employ changes in short lived granulocytes are not using contemporary technology for functional immune system predictive response. At best they are looking "through a glass darkly" and overinterpreting aggregate particle changes as lymphocyte-specific changes, which they probably are not.

The possibility to diagnose delayed allergy (hypersensitivity) with the help of lymphocyte stimulation tests rests on the fact that, in the case of low molecular weight substances (haptens), antigen-specific memory cells are present in patients with allergy symptoms, but not in healthy exposed individuals. Further, since memory cells circulate through the body, the sensitization or allergy is always a systemic phenomenon. The term local allergy, often used in the case of oral mucosal changes, indicates ignorance of modern immunological principles.

The majority of the lymphocytes that operate in cell-mediated reactions are T lymphocytes. T lymphocytes play a key role in the development of all types of allergic and autoimmune disorders. The identification of the antigenic structures (epitopes) involved in allergy and autoimmunity is a hot field in current research. One method of autoimmunity is that metals bind to the sulfhydryl groups on proteins and alter their three-dimensional structure. The immune system recognizes the altered proteins as foreign and an autoimmune process starts, often with condition-specific imbalances in Th1 and Th2 populations of lymphocytes.

TMs and POPs can affect the immune system in several ways. In the oral cavity, high concentrations of metal ions may suppress the immune response and result in immunosuppression. This could explain why the oral mucosa contains only a low number of cells with the capability to present antigen to T lymphocytes. This may also be why mucosal changes adjacent to metal fillings are rarely seen. Higher concentrations of metals can also up-regulate immune reactions (so-called the polyclonal or nonspecific stimulation) and such responses are seen in individuals with intact immunity.

In contrast, in some hereditarily predisposed individuals, TMs may act as haptenic immune reactors. To be able to use the conventional lymphocyte stimulation test for diagnosis of

metal-induced allergy, it was necessary to modify the test in such a way that only the antigen-specific reaction was measured. This was achieved by reducing the concentrations of the metals added to cultures. Since antigen-specific memory cells in the blood are relatively few, the number of lymphocytes in the metal cultures was increased, and the number of other cells that could affect the lymphocyte proliferation negatively was reduced. This version of the lymphocyte stimulation test is called MELISA. Another advanced lymphocyte response assay is the LRA by ELISA/ACT tests system.

In short, MELISA or LRA by ELISA/ACT enables individuals who are immunoreactive to mercury and other metals to be identified. Furthermore, after the removal of amalgam and replacement with nonmetal composites or the systematic reduction in immunoreactive exposures, the lymphocyte stimulation test often reverts to nonreactive. This “resetting” of immune responses typically takes 6-18 months. These changes parallel the decrease in concentrations of mercury inside the neutrophil granulocyte. The dental research in this regard in Sweden is well documented by Hudecek, a capable biological dentist.²¹ After dental amalgam(s) removal, his data showed that 76% of patients reported long-term health improvement, 22% reported unchanged health, and 2% reported a worsening of symptoms.

Recently Lindh and coworkers²¹ reported that at 1 to 2 years after amalgam removal, about one-quarter of patients had completely recovered from their chronic autoimmune or immune dysfunction syndromes; one-half were substantially improved; one-fifth showed no change; and one-twentieth (5%) were worse off than before. This latter group was mostly patients who had improper or premature amalgam removal.

Quantifying Individual Exposure

Evaluation of a person suspected of chronic clinical metal toxicity and/or heavy metal sensitivity or POP burden²² can be based clinically on the following:

1. Determining the body exposure and burden of the TMs, POPs, and the relevant nutritional antitoxic minerals on an appropriately provoked specimen. This is the current state of the art of testing and determination of probable clinical body burden at a sufficient level of precision to warrant clinical management based on the provoked urine quantitative information. In addition, unprovoked urine may be employed as a preprovocative testing screening assessment but is not routinely, clinically necessary.
2. Penicillamine (D-Pen; Cupramine) is an example of a mineral binding or chelating compound that has been standardized for provocative testing and therapeutic monitoring. Penicillamine has been standardized as a challenge agent for cellular toxic and nutritional mineral content. Other chelators are in development, while a variety of selective chelators are currently available, varying with local regulatory practices.
3. The timing of detoxification is best accomplished when host systems for sequestration and rapid elimination of

toxin are facilitated. For example, removal of mercury containing amalgams (if needed) should *follow* a systematic program to enhance dietary intake of detoxifying foods and to reduce the mobilizable burden of TMs or POPs.

Examples of detoxifying foods are garlic, onions, and/or ginger; brassica sprouts; and eggs. Each of them can block uptake and bind (thereby detoxifying) TMs, most POPs, and other sources of biologically active sulfur compounds to accomplish the same effects. Individualized therapy by clinically experienced professionals is needed to guide supplementation, lifestyle changes, and attitudinal healing as needed. These compounds work best when the individual is in a homeostatically balanced lifestyle.

Confirmatory, follow-up testing is encouraged at 3-6 months after the initiation of therapy. In many cases, otherwise unexpected additional toxicants or essential nutritional mineral deficits will be revealed. It is cost effective to engage these elements of comprehensive and integrative care. This reduction in human morbidity can be linked to the reduction of biologically active TMs/POPs and the enhancement of antioxidant, antitoxic stores in the person.

Removing TMs and POPs

As Nriagu and Pacyna remind us, “our toxic metal time bomb’s impact on human health and on our ecosystem has been compared in importance to the total radioactive waste in need of disposal”⁴ or the excess carbon dioxide production associated with enhanced greenhouse gas effects.⁵ This is understandable given the following:

1. An ~1,000 fold increase in TMs and POPs in our environment over the last 1,000 years
2. Over half of the TM and POP burden on the environment has been added within the last century.
3. Bioaccumulation in mammals, including humans, is typically 100,000 to 200,000,000 times that of the environment. This is largely due to most mammals’ ready uptake and impaired innate release (detoxification + excretion) mechanisms.

Now we will look at the practical aspects of identifying nutritional and TM body burden by provocation testing using penicillamine as an example of a validated oral protocol.

Penicillamine Protocol for Determining Toxic and Nutritional Mineral Status in Cells by Noninvasive Provocation into the Urine

Purpose. To determine the body’s burden of mobilizable, potentially toxic metals and the divalent minerals altered when toxins are present.

Table 1 Mineral Value Ranges for Nutritional and Toxic Minerals in 2nd Day 24-hour Urine after Penicillamine Provocation in Healthy Adults*

Mineral element	Reference range (mg/g creatinine)	Reference range (mg/24-h sample)
Nutritional		
Calcium	310-620	400-900
Magnesium	250-550	350-700
Zinc	0.8-1.3	1.1-1.5
Copper	0.04-0.06	0.06-0.08
Iron	0.20-0.30	0.24-0.36
Manganese	0.005-0.007	0.006-0.008
Molybdenum	0.11-0.14	0.13-0.19
Boron	4.1-5.6	5.8-6.7
Chromium	0.19-0.30	0.21-0.33
Cobalt	0.04-0.06	0.05-0.07
Selenium	0.25-0.31	0.24-0.35
Vanadium	0.02-0.03	0.03-0.04
Toxic mineral		
Lead	Reference range ($\mu\text{g/g}$ creatinine) <20	Reference range ($\mu\text{g}/24\text{-h sample}$) <25
Mercury	<7	<9
Arsenic	<120	<175
Nickel	<16	<25
Cadmium	<4	<6

Note: Values lower than the reference range in provoked specimens suggest deficiency of the above-needed essential minerals. Adequacy of supplemental intake to replenish deficits can be monitored by repeat penicillamine provocation every 3 months.

*7.5 mg/kg QID for 3 days, $N = 200$.

Method. A 24-hour urine test during a short noninvasive provocation using oral D-penicillamine (Cupramine, D-Pen, dimethylcysteine, mercaptovaline) or acetyl-penicillamine, prescribed by a physician.

Protocol.

- The average-size adult is prescribed 500 mg of penicillamine or *N*-acetyl-penicillamine with each meal and before bed for 3 days. Generally two capsules of 250 mg each are taken four times a day. This is a total of 2 g/day. The dose is based on 30 mg/kg body weight.
- If weight is under 100 pounds or over 300 pounds, calculation of dose is recommended. For example, a 100-pound adult weighs 45.5 kg. A daily dose of 1,590 mg (~1,500 mg) is recommended. This would most easily be achieved by giving 2×250 mg capsules with breakfast, dinner, and at bedtime (2 capsules TID). By comparison, a 350-pound person weighs 160 kg. At 30 mg/kg, this calculates to a daily dose of 4,800 mg (~4,750 mg). This means taking 5×250 mg capsules with each of three meals plus 4×250 mg capsules at bedtime.
- This short course of penicillamine avoids the rare but important side effects of longer-term therapeutic doses of the drug as discussed in the *Physicians Desk Reference*.²³ Patients should be informed to discontinue taking the medication until otherwise instructed, if there is an adverse effect.
- Begin urine collection on the morning of the second day. This is 24 hours after initiating the penicillamine. Collect all urine in a heavy metal-free container, which is usually provided by the doctor or the laboratory. Urine

must be collected for a full 24-hour cycle. If a urine sample is missed, the collection is incomplete and the protocol should be resumed 1 week later. Urine collected in an incomplete sample may be poured out and the same collection container reused.

- Take the completed urine collection to the laboratory promptly. The total volume is an important part of the information to be sent to the analytic lab. It is desirable, although not necessary, to keep the urine refrigerated during the collection period.
- Analyze the results. Typical penicillamine provocation reference ranges are included in Table 1.
- Please note that a third-day collection cannot be compared with the standardized second-day collection results. Because the specimen is provoked by D-penicillamine, it should not be used for mineral balance studies. The specimen may be used to check kidney function and to analyze for most hormones and neurotransmitter metabolites.
- If substantial total TM tissue burdens are documented, oral pulse therapy (2 days/week) with penicillamine is recommended. Use 7.5 mg/kg QID, on the 2 days each week for 3 months. After 3 months, retest the urine by the penicillamine provocation test to determine residual TM being eliminated as well as comparison of nutritional mineral status. For example, are they assimilating what is being given or do they have enteropathy with consequent reduction in mineral uptake? Do they have particularly high need for particular minerals for their unique metabolic balance state, type, or condition based on functional tests?

Because of short-term effects on other minerals, this specimen should *not* be used for calcium or other mineral balance studies because it is a provoked specimen. The specimen *may* also be used to check kidney function and to analyze for most hormones, neurotransmitter metabolites, etc.

This short course of penicillamine avoids the rare but important side effects of longer-term therapeutic doses of the drug as discussed in the *Physicians Desk Reference*.²³ Of course, if you note any adverse response, discontinue taking the medication until otherwise instructed by your health professional.

Interpretation and Substantiation of Penicillamine Protocol

Each laboratory has an applicable reference range for each mineral assayed. Elevation above the range reported by that laboratory is indicative of increased tissue stores of that heavy metal. Tissue status of nutritional minerals may also be assessed in this way. Typical penicillamine provocation reference ranges are included in Table 1.

Modest Amounts of Provoked TMs

For modest amounts of provoked TMs patients are advised to follow the alkaline way diet. Eighty percent of this diet is comprised of alkaline-forming foods. Alkaline-forming foods add bicarbonate rather than hydrogen ions. One way patients can assess results of their diet is by monitoring urine pH.²⁴

This diet should be combined with sufficient amounts of antioxidants plus minerals (potassium, calcium, magnesium, and zinc as their fully ionized, fully soluble ascorbates, aspartates, citrates, malates, succinates, fumarates, glycinates, or other fully soluble, nonallergenic mineral salts) to displace the TMs. Adequate herbal tea, mineral water, or spring water helps to “wash out” these toxins. Adequate hydration and magnesium are important.

A repeat provocative urine minerals test after 3 to 6 months is recommended to confirm the reduction in available TMs.

For More Than Modest Amounts of Provoked TMs

Use penicillamine twice a week (eg, Monday and Thursday) for 30-60 days at 7.5 mg/kg taken QID (500 mg/QID for most adults) with supplemental calcium, magnesium, and zinc, particularly on the nonpenicillamine days, to replace these minerals (which penicillamine will chelate along with the other divalent [double charged] minerals along with toxic or heavy metals).

Therapeutic doses of antioxidants are beneficial as well as described. These include the following:

1. Buffered ascorbate based on ascorbate calibration to determine physiological ascorbate need.²⁵ Flavonoid/flavanol combinations (eg, a total of 1-30 g daily of quercetin dihydrate and soluble OPC) potentiate the benefits of buffered ascorbate. Their need increases in proportion to buffered ascorbate need as noted in the ascorbate calibration document.
2. Natural vitamins E (mixed tocopherols) 200-1,600 IU/day with tocotrienols (polycosanols) and selenomethionine (300-1500 $\mu\text{g}/\text{day}$).

3. A balanced, high-potency, high-activity B complex including PABA and selenomethionine.
4. A comprehensive mineral supplement is recommended since macro- and micromineral deficits are pervasive. From magnesium and potassium to chromium and vanadium, from manganese and molybdenum to zinc and copper, we can measure the relationships of these key nutritional minerals. Selenomethionine is the most active mineral form for combining with and inactivating TMs.
5. Sulfhydryl-rich foods such as garlic, ginger, and onions; eggs; and brassica vegetables (eg, broccoli, cabbage, etc). Make fresh ginger tea (with raw honey to taste) a staple beverage. A thumb-size piece of fresh ginger, finely chopped, and steeped in hot water for 5 minutes contains over 5,000 μg of toxic mineral-trapping sulfhydryl compounds. Ginger tea may be made up ahead of time and may be drunk cool or cold if preferred.
6. Probiotics (8-20 Bn/day) containing multiple human strains that have been cultured, harvested, and lyophilized (freeze dried) for maximum activity and potency.
7. Carotenoids (eg, 25-100 mg daily of the carotenoid family, including alphacarotene, betacarotene, lutein, cryptoxanthin, and pseudoxanthin) and vitamin D (600-2,400 IU daily) for enhanced cell regulation and resilience. The many roles of vitamin D should be explored.
8. Adequate beneficial, essential fats (eg, 0.5-5 g daily of total omega-3 fatty acids intake), including CLA, DHA, and EPA.

Enhancing antioxidant levels is demonstrated to improve flowing blood in metabolically and hormonally active cells, the blood-brain barrier, and the choroid plexus; the enterocytes in the digestive tract; metabolically active nerve, endocrine, immune, and hepatic cells; sexual function; and skin.

Penicillamine in Clinical Practice

Penicillamine was found to bind copper in the body and safely mobilize it for excretion in the urine (and stool and sweat) of patients with Wilson's disease,²⁶ for which it has remained the treatment of choice for almost half a century. Walshe²⁶ has reported the safe and successful use of penicillamine in pregnant women, infants, the elderly, and the infirm.

In nonhuman species, lead in bone seems to be even more effectively mobilized by penicillamine than lead in soft tissues.^{27,28} However, CaNa_2EDTA is reported to be a more effective lead chelator than penicillamine in vitro in tissue culture.²⁹ Questions have been raised about the safety of using any agent for low-level TM detoxification because some animal studies report that lead may redistribute into soft tissues such as the choroid plexus (where spinal fluid is produced) or in the urine-concentrating loop of Henle in the kidney after CaNa_2EDTA therapy.³⁰ Concerns of this type

have been raised about all oral chelators, although less in regard to penicillamine than any other substance due to the tight bond between toxic minerals and penicillamine.

Lead, Mercury, Arsenic, Cadmium, and Nickel Mobilization by Penicillamine

Clinical benefits of penicillamine are described by Sachs and coworkers³¹ and Vitale and coworkers³² yet not by Marcus³³ (who administered penicillamine while the study subjects continued to live in lead-exposed environs). This may well explain the less dramatic decline in blood lead levels in the Marcus study.

In Chisolm's³⁴ study, children removed from further exposure and treated with penicillamine showed more rapid decline in blood lead levels and in the reversal of hematologic toxicity than the decline in toxicities resulting *solely* from eliminating the lead exposure sources. In contrast, the study by Rogan and coworkers³⁵ did not confirm these findings. This study has been criticized as being flawed in method because the environment was not mitigated for continued TM exposure during the study period.³⁶ In other words, simple use of a chelator is insufficient if the person is left in the intoxicated environment without mitigation.

In addition to lead, penicillamine also mobilizes and facilitates the safer excretion of TMs,³⁷ including mercury,³⁸⁻⁴⁵ arsenic,⁴⁶⁻⁵¹ cadmium,⁵²⁻⁵⁴ and nickel.⁵⁵ Inconsistent reports of efficacy have been published. On balance, these may reflect lack of attention to sufficient reducing substance (ascorbate) to enhance TM mobilization and excretion while maintaining the more effective reduced form of penicillamine rather than its disulfide. An additional factor that reduces TM mobilization is metabolic cellular acidosis. Correction of magnesium buffering deficit aids directly (by displacement) and indirectly (by correcting cellular acidosis) enhanced TM mobilization. Magnesium, as the second most prevalent mineral inside healthy mammalian cells, is a major contributor to cellular buffering and its absence induces cellular metabolic acidosis.²⁴

The toxicity of penicillamine has been described based on its use for several indications in both adults and children. Toxicity of the racemic mixture used years ago to treat chronic arthritis in adults may account for the severity of some of these symptoms and should never be used. In children, nausea and vomiting appear more often at doses exceeding 60 mg/kg per day and may respond to a decrease in dosage.⁴¹ This protocol uses 30 mg/kg doses for just 3 days for provocation.

When given daily and for prolonged periods (which we never recommend), adverse blood and skin effects seem to be idiosyncratic hypersensitivity reactions and are not dose related. Reversible leukopenia or mild thrombocytopenia is reported in less than 10% of children in one study,⁵⁶ but not with similar dosages in two other larger series.⁵⁷ This may have resulted from interaction between penicillamine and pyridoxine (B6).⁵⁸ Supplemental B6 is now routinely recom-

mended as part of penicillamine *therapy* (not provocation). Eosinophilia (defined as >18% eosinophils) has been noted in one-fifth of high-risk children treated daily for an extended duration with the older, less pure preparations of D-pen.^{42,43}

Angioedema, urticaria, or maculopapular eruptions that require discontinuation of drug therapy are reported at a rate of 0.5-1%.⁵⁹ Still less commonly reported reactions are proteinuria, microscopic hematuria, and urinary incontinence.^{46,47} All of these relate to increased tissue permeability due to inhibition of connective tissue cross-links when penicillamine is given on a continuing daily basis and *not* when it is given in the pulsed manner recommended here. All these problems are much less common today due to higher purity of D-pen and improved understanding of its mechanisms of action and how to separate them for clinical benefits.

Distribution in the body of penicillamine is widespread. Like amino acids, such as cysteine, of which penicillamine is the dimethyl analogue as it is also mercaptovaline, it freely moves inside cells, subcellular organelles like the mitochondria, and into deep tissue sites like the brain.⁶⁰⁻⁶⁷

Reactive foods or intestinal irritants such as ferrous sulfate⁶⁸ may reduce the peak level of penicillamine in blood by one-third or more.⁶⁹ Antacids or functional hypochlorhydria⁷⁰ decrease penicillamine absorption by as much as two-thirds.⁷¹ As with all amino acids, peak blood levels are achieved when the amino acid is given on an empty stomach. For provocation and for therapy, mean blood levels are more important than peak blood levels. Thus, taking the penicillamine with food is acceptable. Compliance with this regimen, individually as suggested above, is high.

The recommended dose and duration of therapy with penicillamine have been empirically derived. Doses have ranged from 100 mg/kg per day (in earlier studies) to 20-40 mg/kg/day. Far fewer side effects are reported at the lower dosage range used in provocation and TM mobilization. The duration of the pulse therapy herein recommended is typically on Monday and Thursday for 3 to 6 months, depending on the pretreatment provoked urine TM concentration. When used in this pulsed fashion, penicillamine has become a first line comprehensive care treatment of choice over the several decades of its increasingly widespread use.

Penicillamine has the additional benefit of serving as a source for nitric oxide, which facilitates cellular communication and improved vascular compliance.⁷²

In addition, if substantial total toxic mineral tissue burdens are documented, following the procedure described under the penicillamine protocol is recommended.

Conclusion

Diagnosing and Managing Musculoskeletal Conditions

This section focuses on TMs and POPs with emphasis on their clinical effects, their diagnostic assessment, and their safer detoxification (see Table 2). This is particularly true for health of the body's frame (structure; bones, joints, and extracellular matrix); for metabolic balance (weight, obesity

Table 2 Treatment Guide To Reduce Total Toxic Mineral Tissue Burden and To Enhance Persisting Organic Pollutant Detoxification and Excretion

1. An “alkaline way” energized, high-fiber diet with 80% of what is eaten being alkaline-forming when metabolized. Assessment of first morning urine pH (after a 6 or more hour rest and equilibration period) to assess net acid excess to clinically evaluate metabolic acidosis is recommended.
2. Ginger tea (with raw honey to taste) is rich in beneficial thiols and recommended as a beverage of choice; may be taken warm, cool, or cold.
3. Buffered, fully reduced L-ascorbate intake based on the ascorbate calibration protocol to determine individual ascorbate intake need.
4. Quercetin dihydrate flavonoid and soluble OPC flavanol: 2-10 tablets QID
5. Comprehensive multivitamin/mineral/energizing formula: 2 tablets TID
6. Active human-strain probiotics: 1-6 capsules BID
7. Magnesium as ascorbate, citrate, and glycinate: 2 capsules BID
8. Choline citrate in vegetable glycerol: 1 tsp BID (in juice or water) taken along with Mg Plus Guard to facilitate magnesium uptake while building acetylcholine neurotransmitter, cholinergic bile salts, and buffering/ATP production from citrate and glycerol.
9. The above supplements are given together to gain the cumulative benefit of the following detoxification mechanisms:

Enhanced antioxidant levels result in:

1. Flowing blood in metabolically and hormonally active cells
2. The blood–brain barrier and the choroid plexus
3. The enterocytes in the digestive tract
4. Metabolically active (nerve, endocrine, immune, and hepatic) cells
5. Improved sexual functions
6. Healthy skin look and function

risk, insulin resistance); for managing such “mystery syndromes” as fibromyalgia or myofascial pain syndromes, rhabdomyolysis, or polymyalgia rheumatica; CFIDS or adult failure to thrive syndrome; and for better ergonomic function (athletic ability and injury risk).

With regard to bone health, toxic metals intercalate in bone matrix, decreasing bone strength and falsely elevating apparent bone density. Further, many pollutants adjust pH downward, away from homeostasis and into cellular metabolic acidosis. As a consequence bone is dissolved to buffer and maintain serum acid–alkaline balance even at great metabolic cost to cells and body structures.²⁴

With regard to metabolic balance and weight management (obesity, leptin deficiency), TMs and POPs exacerbate metabolic syndrome (syndrome X, insulin resistance). Since most POPs are fat soluble (lipophilic), as one loses fat weight, toxins are released into the circulation. Increased toxin burden and inflammatory markers have been measured in clinically important amounts. The interaction (synergies) of these toxicants is largely unstudied.

With regard to fibromyalgia and myofascial pain syndromes; rhabdomyolysis (often induced by medications such as HMG–CoA reductase inhibitors/statins) and polymyalgia rheumatica; and CFIDS and adult failure to thrive syndrome, they can be induced and/or exacerbated by TM excess and POPs. TMs and POPs are amongst “nature’s mimics” in that they can bring to substantially greater intensity a wide variety of clinical conditions. This has been well demonstrated in “metal fume fever” in arc welders. Mitochondrial destruction leads to cell apoptosis.

With regard to ergonomics, ergogenics, and sports medicine, peak performance is decreased when muscle mitochondria are damaged or intoxicated (metabolically uncoupled).

This makes the physical body more susceptible to repetitive motion injury or pain from a less than ideal office chair to sit in during the work day. This anti-ergonomic result links commonly with an anti-ergogenic effect. Some sports may even lead to increased toxicity when protective antioxidants and buffering minerals are insufficient to respond to the exercise challenge. An example is lead exposure and bioaccumulation in motocross racers. In other cases, the artist or the draftsman, the computer programmer or the gardener, get exposed to chemicals that inhibit repair.

Our established presumption of safety deserves to give way to a prudent “cautionary principle” that is now evolving in Europe and the Pacific Rim. This means that the burden of proving safety is with the innovator. This is in contrast to our current presumption of safety and aftermarket surveillance to identify risks or toxicities, waiting until the “bodies pile up in the streets” before we take action. The potency, prevalence, and predictable interactions of modern synthetics/toxicant candidates, based on what we now know, has outstripped the ability of aftermarket surveillance to adequately protect public health and safety in a timely and cost-effective manner.

In contrast, natural products, long in traditional use, have established safety profiles so robust that replication in the current pharmaceutical scientific research idiom seems an unnecessary and onerous burden given the goal of regulation being to level the playing field so that the informed consumer could make wiser, more cost- and outcome-effective choices.

The informed clinician now remains the patient’s advocate and safety net. As has often been said, first, we must “do no harm” and, second, we must think of nutrition and detoxification as first line comprehensive care fundamentals of practice.

It is time to turn our attention to the toxic soup of TMs and

POPs that starts with the toxic minerals in our vaccines, continues with the amalgams in our mouth, and continues in the air we breathe, water we drink, and the foods we eat. We can do better. We may have little choice but to do better if we are to survive as healthy, productive, fertile humans.

It is time to start making integrative medicine a higher priority in healthcare practice, policy, and research. The economics are favorable. The patients are seeking and show excellent compliance with these approaches. The health policy community recognizes that prevention and health promotion are superior choices. We contribute that evidence-based best practice guidelines and clinical protocols show documented promise of funding the transition from sick care to healthcare out of savings from sick care costs not incurred due to the judicious and general application of integrative medicine principles.

References

1. <http://www.atsdr.cdc.gov/HEC/CSEM/lead/referencescited.html>
2. Healthbenchmarks.org: Mercury. Available at: <http://www.healthbenchmarks.org/mercury/>
3. Jaffe R, Morris E: Medicine in transition from disease treatment to healthcare. HSC Report 100-14, Sterling, VA, 2000
4. Nriagu JO, Pacyna JM: Quantitative assessment of worldwide contamination of air, water, and soils by toxic metals. *Nature* 333:134-139, 1988
5. Boden TA, Marland G, Andres RJ: CO₂ emission calculations and trends. *Govt Reports Announcements and Index*, Issue 17, 1996
6. MedlinePlus: Mercury. Available at: <http://www.nlm.nih.gov/medlineplus/mercury.html>
7. Toxic Mineral Monographs, ATSDR, CDC, USPHS, 1998-2002
8. Arsenic Monograph, ATSDR, CDC, PHS, GOV, 2000
9. EPA revised arsenic risk assessment. *Chem Eng News*, January 8, 2001
10. ATSDR—ToxFAQs: Cadmium. Available at: <http://www.atsdr.cdc.gov/tfacts5.html>
11. Cohn SL, Goldman L: Preoperative risk evaluation and perioperative management of patients with coronary artery disease. *Med Clin North Am* 87:111-136, 2003
12. Needleman HL: Childhood lead poisoning: The promise and abandonment of primary prevention. *Am J Public Health* 88:1871-1877, 1998
13. <http://www.epa.gov/fedrgstr/EPA-WATER/2001/October/Day-05/w25047.htm>
14. <http://www.iom.edu/CMS/3788/4598/4600.aspx>
15. Lovins A, Lovins H, Hawken P: *Natural Capitalism*. Boston, MA, Brown & Co, 2000
16. Jaffe R: *Clean Your Room*. Report to the Department of Consumer Affairs of the State of California, 1983
17. Walsh WJ: *Biochemical treatment: Medicines for the next century*. *NOHA News* 16:2-4, 1991
18. Maret W, Heffron G, Hill HA, et al: The ATP/metallothioneine interaction: NMR and STM. *Biochemistry* 41:1689-1694, 2002
19. Stejskal VD, Danerslund A, Lindvall A, et al: Metal-specific lymphocytes: Biomarkers of sensitivity in man. *Neuroendocrinol Lett* 20:289-298, 1999
20. Pelletier L, Pasquier R, Hirsch F, et al: In vivo self-reactivity of mononuclear cells to T cells and macrophages exposed to HgCl₂. *Eur J Immunol* 15:460-465, 1985
21. Lindh U, Hudecek R, Danerslund A, et al: Removal of dental amalgam and other metal alloys supported by antioxidant therapy alleviates symptoms and improves quality of life in patients with amalgam-associated ill health. *Neuroendocrinol Lett* 23:459-482, 2002
22. 2nd National Report on Human Exposure to Environmental Chemicals, CDCP, DHHS, January 2003
23. Physicians Desk Reference. Montvale, NJ, Thomson Healthcare, 2005
24. Jaffe R, Brown S: Acid-alkaline balance and its effect on bone health. *Intl J Integrative Med* 4:7-18, 2001
25. Jaffe R: Determination of ascorbate physiologic need by calibration. *Health Studies Collegium Document* 111. (Contact Client Services at 800-525-7372 for reprints)
26. Walshe JM: Penicillamine: A new oral therapy for Wilson's disease. *Am J Med* 21:487-495, 1956
27. Russell JC, Griffin TB, McChesney EW, et al: Metabolism of airborne particulate lead in continuously exposed rats: Effect of penicillamine on mobilization. *Ecotoxicol Environ Saf* 2:49-53, 1978
28. Hammond PB: The effects of D-penicillamine on the tissue distribution and excretion of lead. *Toxicol Appl Pharmacol* 26:241-246, 1973
29. Rosen JF, Markowitz ME: D-Penicillamine: Its actions on lead transport in bone organ culture. *Pediatr Res* 14:330-335, 1980
30. Klaassen CD: Heavy metals and heavy metal antagonists, in Gilman AG, Goodman LS, Rall TW, et al (eds): *The Pharmacological Basis of Therapeutics* (ed 7). New York, NY, MacMillan Publishing, 1985, pp 1605-1627
31. Sachs HK, Blanksma LA, Murray EF, et al: Ambulatory treatment of lead poisoning: report of 1155 cases. *Pediatrics* 46:389-396, 1970
32. Vitale LF, Rosalinas-Bailon A, Folland D, et al: Oral penicillamine therapy for chronic lead poisoning in children. *J Pediatr* 83:1041-1045, 1973
33. Marcus SM: Experience with D-penicillamine in treating lead poisoning. *Vet Hum Toxicol* 24:18-20, 1982
34. Chisolm JJ Jr: Chelation therapy in children with subclinical plumbism. *Pediatrics* 53:441-443, 1974
35. Rogan WJ, Dietrich KN, Ware JH, et al: The effect of chelation therapy with Succimer on neuropsychological development in children exposed to lead: The treatment of lead-exposed children trial group. *N Engl J Med* 344:1421-1426, 2001
36. Shannon M, Woolf A, Binns H, et al: Chelation therapy in children exposed to lead: The treatment of lead-exposed children trial investigators. *N Engl J Med* 345:1212-1213, 2001
37. Chisolm JJ Jr: Poisoning due to heavy metals. *Pediatr Clin North Am* 17:591-615, 1970
38. Greenhouse AH: Heavy metals and the nervous system. *Clin Neuropharmacol* 5:45-92, 1982
39. Satar S, Toprak N, Gokel Y, et al: Intoxication with 100 grams of mercury: A case report and importance of supportive therapy. *Eur J Emerg Med* 8:245-248, 2001
40. Ozuah PO: Mercury poisoning. *Curr Probl Pediatr* 30:91-99, 2000
41. Rosenspire AJ, Bodepudi S, Mathews M, et al: Low levels of ionic mercury modulate protein tyrosine phosphorylation in lymphocytes. *Int J Immunopharmacol* 20:697-707, 1998
42. Finkelstein Y, Vardi J, Kesten MM, et al: The enigma of parkinsonism in chronic borderline mercury intoxication, resolved by challenge with penicillamine. *Neurotoxicology* 17:291-295, 1996
43. Goyer RA, Cherian MG, Jones MM, et al: Role of chelating agents for prevention, intervention, and treatment of exposures to toxic metals. *Environ Health Perspect* 103:1048-1052, 1995
44. Schwartz JG, Snider TE, Montiel MM: Toxicity of a family from vacuumed mercury. *Am J Emerg Med* 10:258-261, 1992
45. Snodgrass W, Sullivan JB Jr, Rumack BH, et al: Mercury poisoning from home gold ore processing: Use of penicillamine and dimercaprol. *JAMA* 246:1929-1931, 1981
46. Cullen NM, Wolf LR, St Clair D: Pediatric arsenic ingestion. *Am J Emerg Med* 13:432-435, 1995
47. Mahajan SK, Aggarwal HK, Wig N, et al: Arsenic induced neuropathy. *J Assoc Physicians India* 40:268-269, 1992
48. Aaseth J: Recent advance in the therapy of metal poisonings with chelating agents. *Hum Toxicol* 2:257-272, 1983
49. Fesmire FM, Schauben JL, Roberge RJ: Survival following massive arsenic ingestion. *Am J Emerg Med* 6:602-606, 1988
50. Watson WA, Veltri JC, Metcalf TJ: Acute arsenic exposure treated with oral D-penicillamine. *Vet Hum Toxicol* 23:164-166, 1981
51. Lyle WH: Penicillamine in metal poisoning. *J Rheumatol Suppl* 7:96-99, 1981
52. Basinger MA, Jones MM, Holscher MA, et al: Antagonists for acute oral

- cadmium chloride intoxication. *J Toxicol Environ Health* 23:77-89, 1988
53. Williams DR, Halstead BW: Chelating agents in medicine. *J Toxicol Clin Toxicol* 19:1081-1115, 1982
54. Freeman HC: Crystal structures of metal-peptide complexes. *Adv Protein Chem* 22:257-424, 1967
55. Shi X, Dalal NS, Kasprzak KS: Generation of free radicals in reactions of Ni(II)-thiol complexes with molecular oxygen and model lipid hydroperoxides. *J Inorg Biochem* 50:211-225, 1993
56. Shannon M, Graef J, Lovejoy FH Jr: Efficacy and toxicity of D-penicillamine in low-level lead poisoning. *J Pediatr* 112:799-804, 1988
57. Bartsocas CS, Grunt JA, Boylen GW Jr et al: Oral D-penicillamine and intramuscular BAL + EDTA in the treatment of lead accumulation. *Acta Paediatr Scand* 60:553-558, 1971 (see also ref. 26)
58. Rothschild B: Pyridoxine deficiency. *Arch Intern Med* 142:840, 1982
59. Holt GA. *Food and Drug Interactions*. Chicago, IL, Precept Press, 1998, p 203
60. Willeit J, Kiechl SG, Birbamer G, et al: Wilson's disease with primary CNS manifestation: Current status in diagnosis and therapy. *Fortschr Neurol Psychiatr* 60:237-245, 1992
61. Meyer BU, Britton TC, Benecke R: Wilson's disease: Normalisation of cortically evoked motor responses with (penicillamine) treatment. *J Neurol* 238:327-330, 1991
62. Maurer K, Ihl R, Dierks T: The topography of P300 in neuropsychiatric pharmacotherapy: III. Cognitive P300 fields in an organic psychosyndrome (Wilson's disease) before and during treatment with D-penicillamine EEG EMG Z Elektroenzephalogr Verwandte Geb 19:62-64, 1988
63. Mizutani N, Maehara M, Negoro T, et al: Serial changes of cranial computerized tomographic findings in Wilson disease during D-penicillamine therapy. *Brain Dev* 5:48-52, 1983
64. Sack G, Lossner J, Bachmann H: Results of electroencephalographic and familial studies in Wilson's disease. *Psychiatr Neurol Med Psychol (Leipz)* 27:455-462, 1975
65. Shimada H, Fukudome S, Kiyozumi M, et al: Further study of effects of chelating agents on excretion of inorganic mercury in rats. *Toxicology* 77:157-169, 1993
66. Bluhm RE, Bobbitt RG, Welch LW, et al: Elemental mercury vapour toxicity, treatment, and prognosis after acute, intensive exposure in chloralkali plant workers: Part I. History, neuropsychological findings and chelator effects. *Hum Exp Toxicol* 11:201-210, 1992
67. Kern F, Roberts N, Ostlere L, et al: Ammoniated mercury ointment as a cause of peripheral neuropathy. *Dermatologica* 183:280-282, 1991
68. Harkness JAL, Blake DR: Penicillamine nephropathy and iron. *Lancet* 2:1368-1369, 1982
69. Osman MA, Patel RB, Schuna A, et al: Reduction in oral penicillamine absorption by food, antacid, and ferrous sulfate. *Clin Pharmacol Ther* 33:465-470, 1983
70. Threlkeld DS (ed): *Miscellaneous products, penicillamine*, in *Facts and Comparisons Drug Information*. St. Louis, MO, JB Lippincott, 1996, pp 714-716b
71. Ifan A, Welling PG: Pharmacokinetics of oral 500 mg penicillamine: Effect of antacids on absorption. *Biopharm Drug Dispos* 7:401-405, 1986
72. Stefano GB, Hartman A, Bilfinger TV, et al: Presence of the mu3 opiate receptor in endothelial cells. Coupling to nitric oxide production and vasodilation. *J Biol Chem* 270:30290-30293, 1995